Formation of 2-oxa- or 2-azabicyclo[3.3.0]octa-3,7-diene by a novel tandem intramolecular photo-cyclization of 2,4,6-tris(phenylthio)hepta-2,4,6-trienal derivatives

Mitsuhiro Yoshimatsu,*a Satoshi Gotoh,a Genzoh Tanabeb and Osamu Muraokab

^a Department of Chemistry, Faculty of Education, Gifu University, Yanagido, Gifu 501-1193, Japan. E-mail: yoshimae@gumail.cc.gifu-u.ac.jp

^b Kinki University, Faculty of Pharmaceutical Sciences, 3-4-1, Kowakae, Higashi-osaka, Osaka 577-8502, Japan

Received (in Cambridge, UK) 3rd March 1999, Accepted 12th March 1999

The photo-reactions of 2,4,6-tris(phenylthio)hepta-2,4,6-trienal 1 and its 2,4-dinitrophenylhydrazone 5 gave the 2-oxaor 2-azabicyclo[3.3.0]octa-3,7-dienes 2 and 9, respectively, *via* a photo-induced intramolecular tandem cyclization reaction.

The Paterno–Büchi reaction has become increasingly familiar to chemists engaged in investigation of its mechanism or synthetic applications.¹ The widespread advances in this area of synthetic organic chemistry have led to convenient methodology for the preparation of the oxetane-containing biologically active materials.² Intramolecular photo-reactions of the viny-loxy ω -carbonyl compounds and their furan derivatives usually provides the fused oxetane ring compounds,³ which lead to other useful compounds by successive transformations.⁴

Recently, we reported the tandem α -(thio)- or α -(seleno)formylolefination reactions of the aldehydes.⁵ Three cycles of the olefination process with pivalaldehyde stereoselectively afforded (2*E*, 4*E*, 6*E*)-8,8-dimethyl-2,4,6-tris(phenylthio)nona-2,4,6-trienal **1**, which contains three conjugated vinyl sufide moieties. This trienal **1** was found to be light-sensitive like the retinal chromophores.⁶ This result prompted us to investigate the reactivities of the trienal under photo-irradiation conditions; surprisingly, a new ring system, oxabicyclo[3.3.0]octa-3,7-diene, was formed *via* an unusual photo-cyclization. Here we report the unprecedented intramolecular tandem cyclization reaction of trienal **1**.

Irradiation of an Ar saturated solution of 1 in MeCN (1.0 \times 10⁻³ mol l⁻¹) using a 40 W fluorescent daylight lamp afforded (1R*, 5R*, 6S*)-6-tert-butyl-1,4,7-tris(phenylthio)-2-oxabicyclo[3.3.0]octa-3.7-diene 2⁷ in 54% yield as shown in Scheme 1. The structure of 2 followed from the fact that the IR spectrum showed no carbonyl absorption, while the ¹H NMR spectrum showed two methine protons at δ 2.82 (t, J 2) and 3.40 (t, J 2) and two olefinic protons at δ 5.07 (s) and 6.50 (d, J 2). The stereochemistry of oxabicyclo compound 2 was determined as $(1R^*, 5R^*, 6S^*)$ by single crystal X-ray analysis; the ORTEP drawing of 2 is shown in Fig. 1.8 Furthermore, a few fundamental reactions of 2 were examined using oxidizing agents. MCPBA oxidation of 2 gave the corresponding sulfone 3 in 78% yield (Scheme 2). The reaction with Pd-C in toluene under reflux conditions afforded the ring-opened cyclopentadiene 4 in 40% yield. The structure of 4 was determined from the IR spectrum, which showed the carbonyl absorption at v1650 cm⁻¹, the ¹H NMR spectrum, which exhibited two doublets at δ 6.54 (d, J7) (due to the olefinic proton) and 11.13



Scheme 1 Reagents and conditions: i, hv (500-600 nm), MeCN.



Scheme 2 Reagents and conditions: i, MCPBA, ClCH₂CH₂Cl; ii, Pd–C, toluene.

(d, *J* 7) (due to the aldehyde proton), and the mass spectrum, which showed a molecular ion peak ($C_{28}H_{22}OS$) at m/z 378. The stereochemistry of **4** was determined as *Z* by an NOE experiment. Irradiation of the *tert*-butyl protons at δ 1.22 increased the intensity of the olefinic proton at δ 6.54 (4%).

Next, we examined the photo-reaction of the trienal hydrazone **5**, irradiating at 400–500 nm in benzene to give 2-azabicyclo[3.3.0]octa-3,7-diene **6** in 41% yield (Scheme 3).⁹



Scheme 3 Reagents and conditions: i, hv (400–500 nm), benzene; ii, AIBN, benzene, reflux.



Recently, we reported that the 2,4-dinitrophenylhydrazones reacted with AIBN and underwent dearylamination.¹⁰ Treatment of the hydrazone **6** with AIBN in benzene afforded 2-azabicyclo[3.3.0]octa-3,5,7-triene **7** in good yield.

A plausible mechanism for formation of the photo-reaction products is shown in Scheme 4. Intermolecular [2 + 2] photocycloadditions have been intensively investigated and have indicated that the reaction intermediate is a flexible 1,4-biradical.¹¹ Our intramolecular cyclization would also proceed *via* triplet biradical intermediates and cyclize in 5-*exo* mode to give the 1,4-biradical **9**. Isomerization of **9** affords a 1,5-biradical **10**. Tandem cyclization leads to the bicyclo-[3.3.0]octa-3,7-diene **11**.

In conclusion, 2,4,6-tris(phenylthio)hepta-2,4,6-trienal derivatives are highly sensitive to light and their photo-reactions were found to afford the 2-oxa- or 2-azabicyclic compounds *via* a unique tandem cyclization reaction.

Notes and references

- J. A. Porco and S. L. Schreiber, *Comprehensive Organic Synthesis*, ed. B. M. Trost, I. Fleming and L. A. Paquette, Pergamon, New York, 1991, vol. 5, p. 151.
- S. S. Bhagwat, P. R. Hamann, W. C. Still, S. Bunting and F. A. Fitzpatrick, J. Am. Chem. Soc., 1985, **107**, 6372; S. S. Bhagwat, P. R. Hamann, W. C. Still, S. Bunting and F. A. Fitzpatrick, Nature, 1985, **315**, 511; D. W. Norbeck and J. B. Kramer, J. Am. Chem. Soc., 1988, **110**, 7217; Y. Kawahata, S. Takatsuto, N. Ikekawa, M. Murata and S. Omura, Chem. Pharm. Bull., 1986, **34**, 3102.
- M. Sakamoto, Y. Omote and H. Aoyama, J. Org. Chem., 1984, 49, 396;
 G. Pattenden and A. M. Birch, J. Chem. Soc., Chem. Commun., 1980,

1195; M. J. Begley, A. M. Birch and G. Pattenden, J. Chem. Soc., Chem. Commun., 1979, 235.

- 4 H. A. J. Carless, J. Beanland and S. Mwesigye-Kibonde, *Tetrahedron Lett.*, 1987, 28, 5933; M Riediker and J. Schwartz, *J. Am. Chem. Soc.*, 1982, 104, 5842; K. Maruyama and Y. Kubo, *J. Org. Chem.*, 1977, 42, 3215; P. H. Mazzocchi, L. Klingler, M. Edwards, P. Wilson and D. Shook, *Tetrahedron Lett.*, 1983, 24, 143.
- 5 M. Yoshimatsu, K. Oguri, K. Ikeda and S. Gotoh, *J. Org. Chem.*, 1998, **63**, 4475.
- 6 Q. Tan and K. Nakanishi, J. Am. Chem. Soc., 1998, 120, 12357 and references cited therein.
- 7 Selected data for **2**: v_{max}/cm^{-1} 1130 (ether); δ_{H} 0.91 (9H, s, Me × 3), 2.82 (1H, t, J 2, 6-H), 3.40 (1H, t, J 2, 5-H), 5.07 (1H, s, 8-H), 6.50 (1H, d, J 2, 3-H), 6.55–6.59 (2H, m, ArH), 7.01–7.07 (3H, m, ArH), 7.36–7.42 (5H, m, ArH), 7.46–7.51 (3H, m, ArH), 7.60–7.62 (2H, m, ArH); δ_{C} 28.60 (q × 3), 34.54 (s), 54.53 (d), 64.28 (d), 107.23 (s), 107.77 (s), 123.29 (d), 125.03 (d × 2), 126.16 (d), 128.89 (d × 2), 129.00 (d), 129.07 (d), 129.15 (d × 2), 129.57 (d × 2), 131.19 (s × 2), 132.13 (s), 132.46 (d × 2), 135.95 (d × 2), 150.10 (d), 150.65 (s); m/z488 (M⁺) (Calc. for C₂₉H₂₈OS₃; C, 71.27; H, 5.77. Found: C, 71.12; H, 5.77%).
- 8 *Crystal data* for **2**: $C_{29}H_{28}OS_3 M = 488.72$, monoclinic, a = 9.007(2), b = 26.879(6), c = 10.945(2) Å, $\beta = 99.34(2)^{\circ}$, V = 2614.7(9) Å³, T = 298 K, space group $P2_1/n$, Z = 4, μ (Cu-K α) = 2.728 mm⁻¹, $D_c = 1.214$ Mg m⁻³, 4284 reflections collected (Rigaku AFCSR diffractometer) of which 4096 were unique ($R_{int} = 0.030$) and 2720 were observed [$I > 3.00\sigma(I)$]. Solved by direct methods (MITHRIL90 and DIRDIF) and refined by full-matrix least-squares (teXsan) on *F* of all unique data to R = 0.056 (observed data), wR = 0.070 (all data). CCDC 182/1222. See http://www.rsc.org/suppdata/cc/1999/909/ for crystal-lograpic files in .cif format.
- 9 Selected data for **6**: v_{max}/cm^{-1} 3330 (NH), 1620, 1340 (NO₂); $\delta_{\rm H}$ 1.28 (9H, s, Me × 3), 4.47 (1H, s, 6-H), 6.29 (1H, s, 8-H), 6.47 (1H, s, 5-H), 6.57 (1H, s, 3-H), 6.74 (1H, d, J 10, ArH), 7.07–7.40 (15H, m, ArH), 8.13 (1H, dd, J 2 and 10 ArH), 9.02 (1H, br d, J 2, ArH), 10.06 (1H, s, NH); $\delta_{\rm C}$ 30.39 (q × 3), 33.85 (s), 67.58 (d), 101.02 (s), 114.20 (d), 121.73 (s), 123.65 (d), 123.71 (d), 125.69 (d), 126.19 (d), 126.42 (d), 126.66 (d × 2), 128.23 (d), 128.45 (d × 2), 128.98 (d), 129.24 (d), 129.30 (d), 129.58 (s), 130.52 (d), 138.33 (s), 138.68 (s), 130.274 (s), 133.92 (d), 133.98 (d), 134.08 (s), 138.33 (s), 138.68 (s), 140.31 (d), 147.70 (s), 152.78 (d); m/z 429 [M⁺ (Bu^t + (O₂N)₂C₆H₃NH)] (Calc. for C₃₅H₃₂N₄O₄S₃: C, 62.85; H, 4.82; N, 8.38. Found: C, 62.71; H, 5.23; N, 7.96%).
- 10 M. Yoshimatsu, S. Gotoh, K. Ikeda and M. Komori, J. Org. Chem., 1998, 63, 6619.
- 11 A. G. Griesbeck, S. Buhr, M. Fiege, H. Schmickler and J. Lex, J. Org. Chem., 1998, 63, 3847.

Communication 9/01707K